

<i>Storage form</i>	<i>Energy density</i>		<i>Cycle efficiency</i>
	<i>kJ kg⁻¹</i>	<i>MJ m⁻³</i>	
<i>Conventional fuels</i>			
Crude oil	42 000	37 000	
Coal	32 000	42 000	
Dry wood	12 500 ^a	10 000	
<i>Synthetic fuels</i>			
Hydrogen, gas	120 000	10	0.4–0.6
Hydrogen, liquid	120 000	8 700	
Hydrogen, metal hydride	2 000–9 000	5 000–15 000	
Methanol	21 000	17 000	
Ethanol	28 000	22 000	
<i>Thermal – low quality</i>			
Water, 100°C → 40°C	250	250	
Rocks, 100°C → 40°C	40–50	100–140	
Iron, 100°C → 40°C	~30	~230	
<i>Thermal – high quality</i>			
Rocks, e.g. 400°C → 200°C	~160	~430	
Iron, e.g. 400°C → 200°C	~100	~800	
Inorganic salts, heat of fusion > 300°C	> 300	> 300	
<i>Mechanical</i>			
Pumped hydro, 100 m head	1	1	0.65–0.8
Compressed air		~15	0.4–0.5
Flywheels, steel	30–120	240–950	
Flywheels, advanced	> 200	> 100	~0.95
<i>Electrochemical</i>			
Lead–acid	40–140	100–900	0.7–0.8
Nickel–cadmium	~350	~350	varying
Lithium ion (other advanced batteries)	700 (> 400)	1400 (> 300)	0.7 (> 0.8)
<i>Superconducting</i>		~100	~0.85

Table 5.5. Energy density by weight and volume for various storage forms, based on measured data or expectations for practical applications. For the storage forms aimed at storing and regenerating high-quality energy (electricity), cycle efficiencies are also indicated. Hydrogen gas density is quoted at ambient pressure and temperature. For compressed air energy storage, both electricity and heat inputs are included on equal terms in estimating the cycle efficiency (with use of Jensen and Sørensen, 1984).

^a Oven-dry wood may reach values up to 20 000 kJ kg⁻¹.